

Development of a Classical Potential for PbTiO_3

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The use of ferroelectric perovskite oxides in a variety of technological applications has prompted extensive investigations of their structure and dynamics. First-principles density functional theory (DFT) calculations have played an important role in enhancing microscopic understanding of the relationships between composition, structure and properties [1–3]. Despite the success of first-principles methods, the great computational expense and restriction to studying zero-temperature properties have driven the development of more efficient atomistic and effective Hamiltonian potentials suitable for large-scale molecular dynamics (MD) simulations.

We present a modified bond-valence model of PbTiO_3 based on the principles of bond-valence [4] and bond-valence vector conservation [5]. The relationship between bond-valence energy and bond-order potential is derived analytically in the framework of a tight-binding model [6]. A new energy term, bond-valence vector energy, is introduced into the atomistic model and the potential parameters are re-optimized. The new model potential can be applied to both canonical ensemble (*NVT*) and isobaric-isothermal ensemble (*NPT*) molecular dynamics (MD) simulations. This model reproduces the experimental phase transition temperature in *NVT* MD simulations and also exhibits the experimental sequence of temperature-driven and pressure-driven phase transitions in *NPT* simulations. We expect that this improved bond-valence model can be applied to a broad range of inorganic materials.

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- [1] R. E. Cohen, *Nature* **358**, 136 (1992).
- [2] W. Zhong, R. D. Kingsmith, and D. Vanderbilt, *Phys. Rev. Lett.* **72**, 3618 (1994).
- [3] I. Grinberg and A. M. Rappe, *Phys. Rev. B* **70**, 220101 (2004).
- [4] I. D. Brown, *Chem. Rev.* **109**, 6858 (2009).
- [5] M. A. Harvey, S. Baggio, and R. Baggio, *Acta Crystallogr. B* **62**, 1038 (2006).
- [6] L. Hansen, P. Stoltze, K. W. Jacobsen, and J. K. Nørskov, *Phys. Rev. B* **44**, 6523 (1991).